Answer **all** the questions in the spaces provided.

1 The figure below shows the graph obtained in an experiment where a solid substance **X** was heated from room temperature to 200°C.



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(c) At the start of the experiment there is 4.8 g of magnesium. Calculate the amount of magnesium in moles.

Amount of magnesium = mol [1]

[2]

3 Cars which run on petrol as their fuel are the main source of air pollution today. Some of the substances found in the exhaust gases of a car are shown below.



- (a) From the diagram above, name the gas that is:

 - (ii) produced in the engine of the car as a result of the high temperature in the engine.

.....[1]

- (b) Carbon monoxide is a gas found in the exhaust gases of the car. It is fatal to human beings when large quantities are inhaled.
 - (i) Briefly describe how carbon monoxide is produced in the car engine.

......[1]

(ii) Explain how carbon monoxide harms the human body when it is inhaled.

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.....[1]

4 Substance **X** is a mixture of dyes that was recently developed by an ice-cream company who are planning to use it in ice-creams.

A paper chromatography was performed to check if substance **X** is safe for consumption. Dyes **A**, **B**, **C** and **D** are known harmful substances. Water was used as the solvent. The diagram below shows the resulting chromatogram.



(a) Is substance **X** safe to be consumed? Explain your answer.

......[1]

(b) Dye **D** did not separate into its components. Suggest why this is so.

......[1]

(c) What will happen if the starting line is below the water level?

.....[1]

Section B [16 marks]

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